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has a definite composition. Ozone is a chemical that is made up of three particles of oxygen. Ozone

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Chapter 3 Guided Notes 3.1 Measurements and Their ...

GUIDED PRACTICE PROBLEM 3 (page 327) 3. Balance each equation. a. $\text{AgNO}_3 + \text{H}_2\text{S} \rightarrow \text{Ag}_2\text{S} + \text{HNO}_3$ b. $\text{Zn}(\text{OH})_2 + \text{H}_3\text{PO}_4 \rightarrow \text{Zn}_3(\text{PO}_4)_2 + \text{H}_2\text{O}$ a. $\text{AgNO}_3 + \text{H}_2\text{S} \rightarrow \text{Ag}_2\text{S} + \text{HNO}_3$ Analyze Step 1. Identify the relevant concepts. Count the number of atoms of each element on both sides of the skeleton equation.

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multi-step problems. (pg.85) 41. Do practice problem 34 on page 85. 34) 42. You can use dimensional analysis to convert just as easy as converting single units. (pg. 86) 43. Do practice problem 36 on page 86. 36) 3.4 Density 44. Look at Figure 3.13. What is the same about the three cubes? What two things are different?

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321-329)

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